

Ten years of reactive transport modeling and simulation

Jocelyne Erhel, Michel Kern

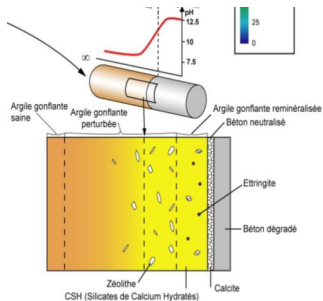
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Journées Scientifiques du GNR MoMaS
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Motivations

Importance of **chemical phenomena** in various applications

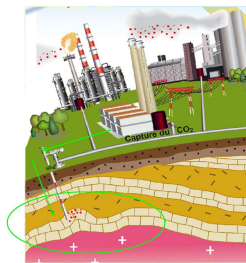
- Alteration chimique des composants du stockage
- Séquestration du CO₂



G. Pépin (Andra)

Carbon
Capture
and
Storage

Aquifères Salins
Gisements Déplétés
Veines de Charbon



Source BRGM / IFP

Capture
↓
Transport
↓
Stockage

A. Michel (IFP)

- 1 Models
- 2 Formulation and numerical methods
- 3 Examples

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Mobile species

$$\omega \partial_t \mathbf{c}_i + \underbrace{\nabla \cdot (\mathbf{u} \mathbf{c}_i - \mathbf{D} \nabla \mathbf{c}_i)}_{L(\mathbf{c})} = \sum_{j=1}^{N_R} \nu_{ij} R_j(\mathbf{c}_1, \dots, \mathbf{c}_{N_S}), \quad i = 1, \dots, N_S$$

- \mathbf{c}_i : concentration of i th species [mol/l]
- \mathbf{D} Dispersion – diffusion tensor [m^2/s]
- R_j reaction term for j th reaction
- ω : porosity [–]
- \mathbf{u} Darcy velocity [m/s]
- ν_{ij} stoichiometric coeff.

Transport with chemical reactions

Mobile species

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Condense transport solver, one time step

$$\mathbf{c}(t + \Delta t) = \Psi_T(R, \mathbf{c}(t))$$

Mass balance for immobile species

$$\rho_s \partial_t \bar{\mathbf{c}}_i = \sum_{j=1}^{N_R} \nu_{ij} R_j(\mathbf{c}_1, \dots, \mathbf{c}_{N_S}, \bar{\mathbf{c}}_1, \dots, \bar{\mathbf{c}}_{N_S}), \quad i = 1, \dots, N_S$$

Modeling general equilibrium models

$$\sum_{j=1}^{N_S + \bar{N}_S} \nu_{ij} Y_j \rightleftharpoons 0, \quad i = 1, \dots, N_r \quad N_S + \bar{N}_S \text{ species}, N_r \text{ reactions.}$$

Mass action law $\nu \log \begin{pmatrix} c \\ \bar{c} \end{pmatrix} + \log K = 0$

Mass conservation $\mathbf{v}^T \begin{pmatrix} c \\ \bar{c} \end{pmatrix} = \begin{pmatrix} T \\ W \end{pmatrix}$

System of non-linear equations
 T known from transport, W
imposed

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Mineral reactions

Dissolution of solid, precipitation of aqueous species. Reactions with threshold : **which** species appear unknown a priori.

$$\text{Solubility product } \Pi = \log K_p + S_p \log c, \quad \begin{cases} p = 0 & \text{if } \Pi < 0 \\ \Pi = 0 & \text{otherwise} \end{cases}$$

Numerical solution of nonlinear problem

Take concentration **logarithms** as main unknowns

Use **globalized** Newton's method (line search, trust region).

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Methods for minerals

- Standard procedure : **combinatorial** search
- Reformulate as **complementarity** problem
- **Interior point** algorithm (Saaf et al. ('96), MK (05))
- Also **semi-smooth** Newton (Kräutle)

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Role of chemical model

Given totals T (and W , known), split into mobile (C) and immobile (F) total concentrations

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Chemistry solver

$$H \begin{pmatrix} \log c \\ \log \bar{c} \end{pmatrix} = \begin{pmatrix} T \\ W \end{pmatrix}$$

$$F = \Psi_C(T, W)$$

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Formulations and solution strategies

Eliminate **unkown** equilibrium reaction rates by introducing **mobile** and **immobile** totals.

$$\phi \partial_t C + \partial_t F + LC = 0 \text{ with } F = \Psi_c(C + F)$$

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Others

- Nonlinear conjugate gradient (Bouillard, Herbin, Montarnal)
- Elimination technique (Knabner, Kräutle, Hoffmann)

De Dieuleveult, JE, MK (JCP '09)

CC formulation, explicit chemistry

$$\begin{cases} \phi \partial_t C + \partial_t F + LC = 0 \\ H \begin{pmatrix} \log c \\ \log \bar{c} \end{pmatrix} - \begin{pmatrix} C + F \\ W \end{pmatrix} = 0 \\ F - F \begin{pmatrix} \log c \\ \log \bar{c} \end{pmatrix} = 0. \end{cases}$$

- + **Explicit** Jacobian
- + Chemistry function, no chemical **solve**
- – **Intrusive** approach (chemistry not a black box)
- – **Precipitation** not easy to include (semi-smooth Newton OK)

Coupled system is index 1 DAE

$$K \frac{dy}{dt} + f(y) = 0$$

Use standard DAE software

C. de Dieuleveult (Andra thesis),

A global method from the fixed–point formulation (1)

Discrete non-linear system

$$\begin{cases} C^{n+1} = \Psi_T \left(\phi \frac{F^n - F^{n+1}}{\Delta t}, C^n \right) \\ F^{n+1} = \Psi_C(C^{n+1} + F^{n+1}) \end{cases}$$

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Residual computation

- 1 Apply Ψ_T : solve **transport** for each species,
- 2 Apply Ψ_C : solve **chemistry** for each grid cell.

- + **Non-intrusive** approach
- + **Precipitation** can be included
- – One chemical equilibrium **solve** for each function evaluation

Solution by **Newton–Krylov** method

- Solve the linear system by an **iterative** method (GMRES)
- Requires only jacobian matrix by vector products, Jacobian not stored
- Keep transport and chemistry as black–boxes (up to Jacobian computation)

Used for CFD, shallow water, radiative transfer and reactive transport
(Hammond, Valocchi, Lichtner, Adv. Wat. Res. 05)

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Inexact Newton

- **Approximation** of the Newton's direction $\|f'(x_k)d + f(x_k)\| \leq \eta \|f(x_k)\|$
- Choice of **the forcing** term $\eta = \gamma \|f(x_k)\|^2 / \|f(x_{k-1})\|^2$ (Kelley, Eisenstat and Walker)
 - Keep superlinear convergence (locally)
 - Avoid oversolving the linear system

L. Amir's thesis, Amir, MK (Comp. Geosci. 09)

Preconditioning

- Essential for good linear performance
- Difficult for matrix free formulation
- Simplified, 1 species model, with explicit sorption (with A. Taakili)

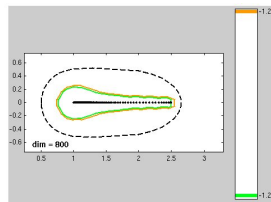
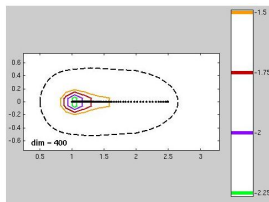
Algebraic elimination of mobile conc. equivalent to Schur complement of block Gauss–Seidel precondition.

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Can show eigenvalues of preconditioned op. bounded away from 0, independent of h , but convergence of GMRES **not** determined by eigenvalues
Field of values analysis ?



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The Reactive transport benchmark

The setup

- Designed by **J. Carrayrou**, M. Kern, P. Knabner
- Concentrate on **numerical** difficulties : simple geometry, « abstract » chemistry
- 3 levels of chemistry (sorption, equilibrium minerals, kinetics)

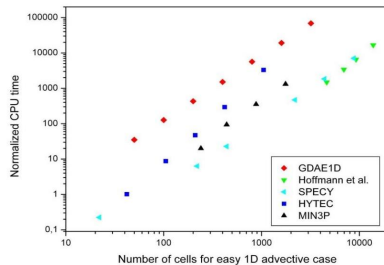
The benchmark

- Results from 6 groups, awards to 4 groups
- International workshop (Ph. Ackerer, Strasbourg, Jan. 2008)
- Special issue in « Computational Geosciences » (ed. Ph. Ackerer), 6 papers + intro + synthesis

1D benchmark results

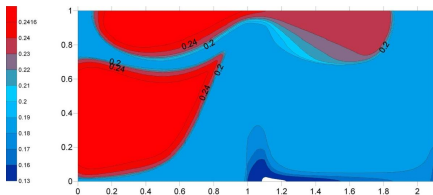
	Loc. of peak	S conc.
GDAE1D	0.0175	0.966
NK	0.0167	0.742
Erlangen	0.0167	0.852
Specy	0.0158	0.968
HYTEC	0.0170	0.286
MIN3P	0.0175	0.725
Reference	0.0173	0.985

GDAE1D more accurate, but slower

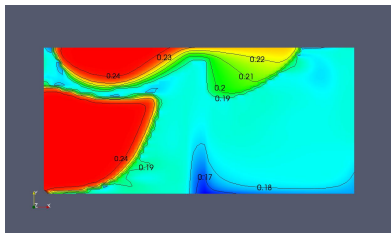


2D benchmark results

One species among 13 at time $t=1000$



Erlangen result with very fine mesh



GDAE result with coarse mesh
Numerical dispersion due to the coarse
mesh but accurate results

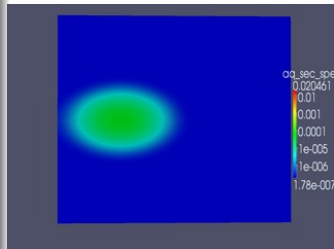
2D Andra test case

Chemical description

- Injection of alkaline water (NaOH) into a porous medium containing quartz (SiO₂)
- Dissolution of quartz : $\text{H}_4\text{SiO}_4 \rightleftharpoons \text{SiO}_2 + 2 \text{H}_2\text{O}$
- Aqueous reactions : $\text{H}_4\text{SiO}_4 \rightleftharpoons \text{H}_3\text{SiO}_4^- + \text{H}^+$, $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
- Sodium is a tracer

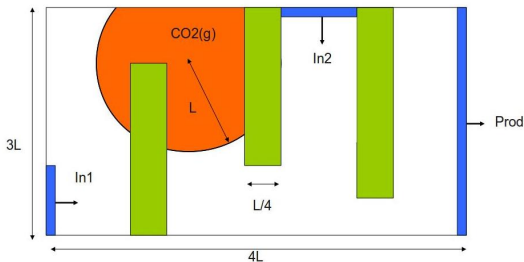
Geometry and transport conditions

- Rectangular domain of size 5m x 3.5m
- Injection at time $t = 0$ of NaOH at point (1, 1.75)
- Advection ($v = 5.710^{-7} \text{ m/s}$) and dispersion
- Duration 30 days



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Minimal chemical system that still "looks" realistic for CO₂ storage

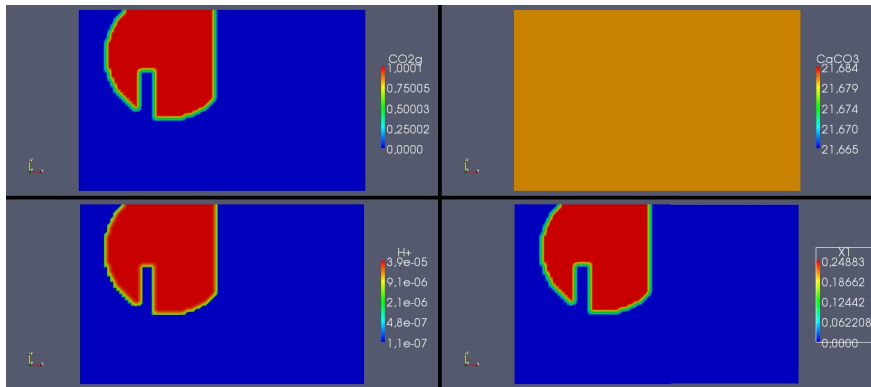


Dissolution of CO₂ in water, dissolution of calcite. Gas assumed **immobile** (capillary trapping), decouples flow from reactive transport.

Chemical system

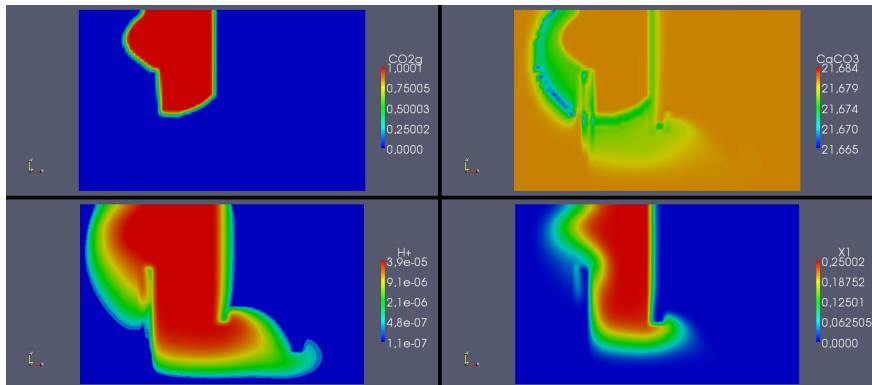
- $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$ water dissociation
- $\text{H}_2\text{O} + \text{CO}_2(\text{aq}) \rightleftharpoons \text{HCO}_3^- + \text{H}^+$ dissociation of aqueous CO₂
- $\text{CO}_2(\text{g}) \rightleftharpoons \text{CO}_2(\text{aq})$ gas dissolution
- $\text{CaCO}_3 + \text{H}^+ \rightleftharpoons \text{Ca}_2^+ + \text{HCO}_3^-$ Dissolution of calcite

Evolution of concentrations (Post-doc B. Gueslin)



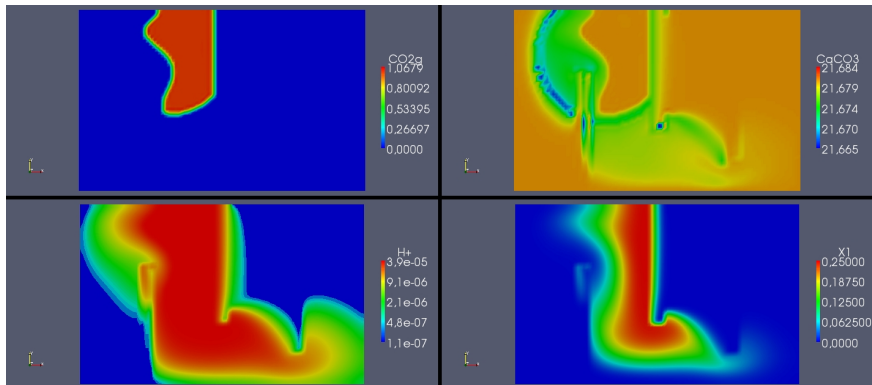
$t = 0$

Evolution of concentrations (Post-doc B. Gueslin)



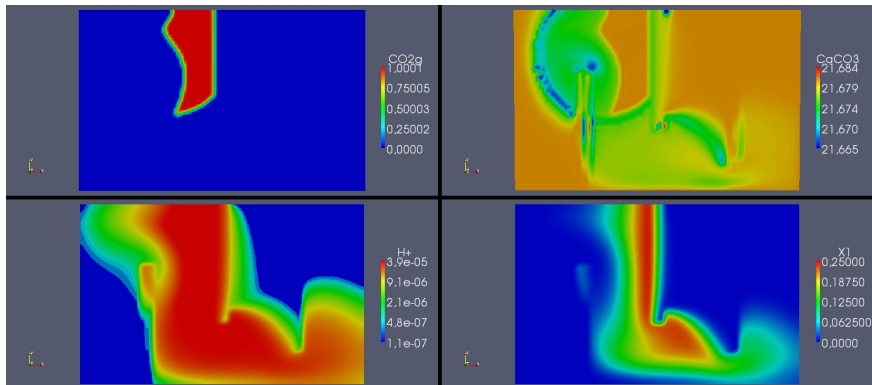
$t = 400$ years

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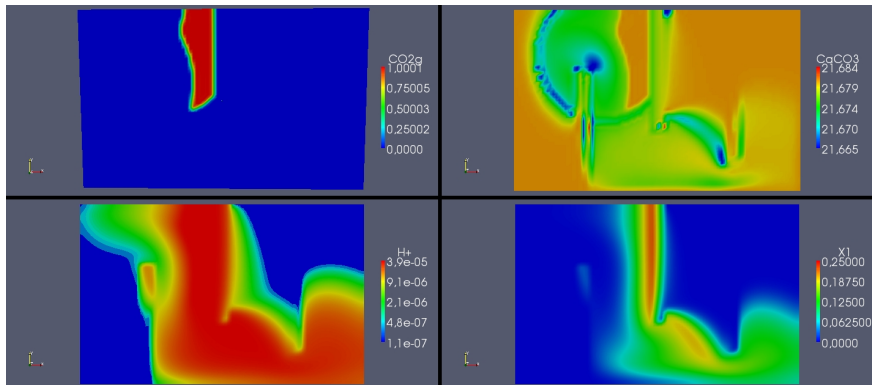
$t = 800$ years

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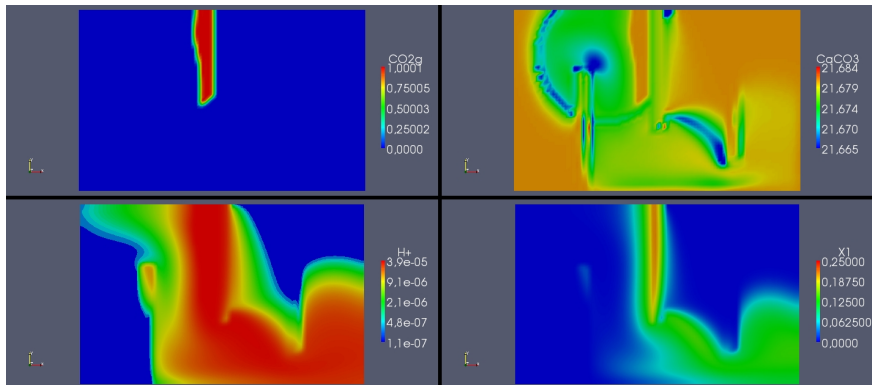
$t = 1200$ years

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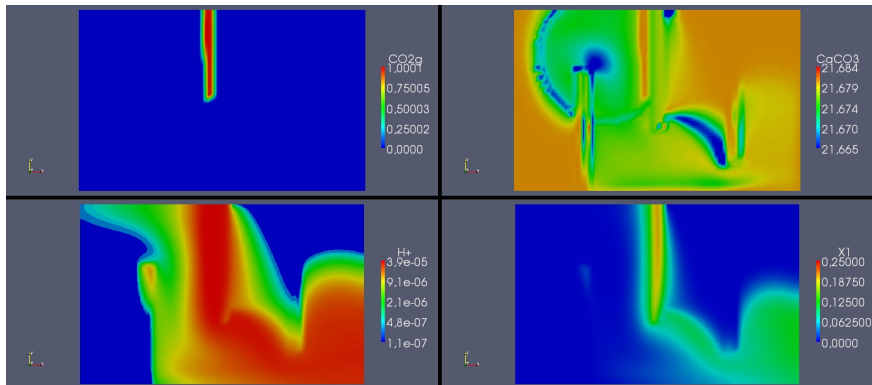
$t = 1600$ years

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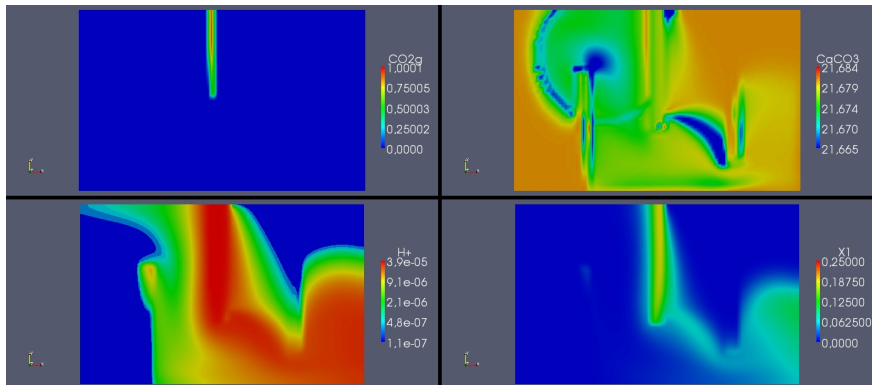
$t = 2000$ years

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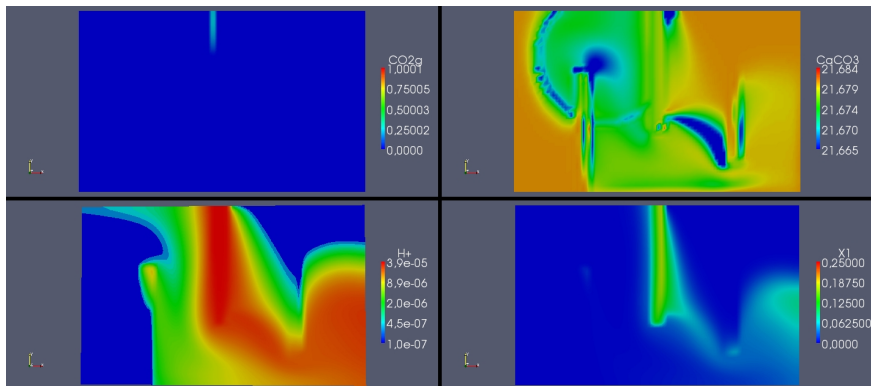
$t = 2400$ years

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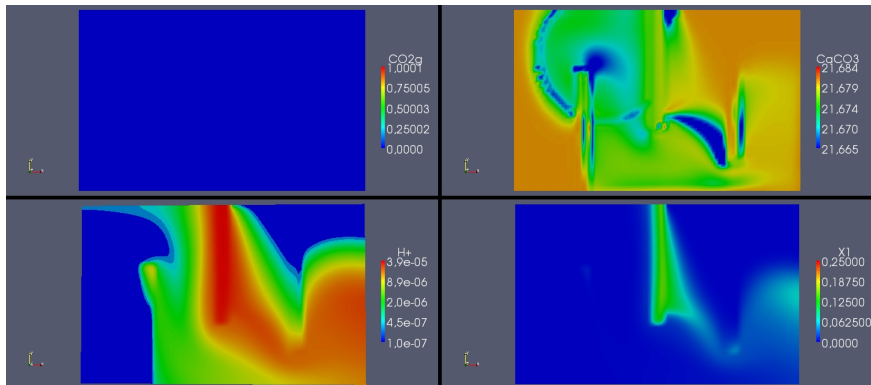
$t = 2800$ years

Evolution of concentrations (Post-doc B. Gueslin)



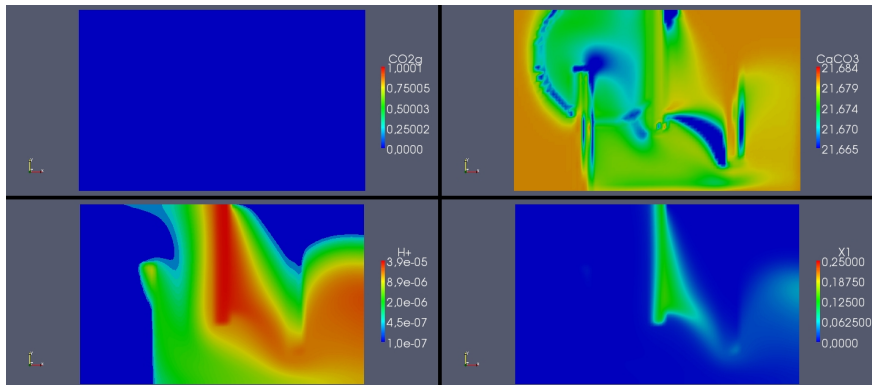
$t = 3200$ years

Evolution of concentrations (Post-doc B. Gueslin)



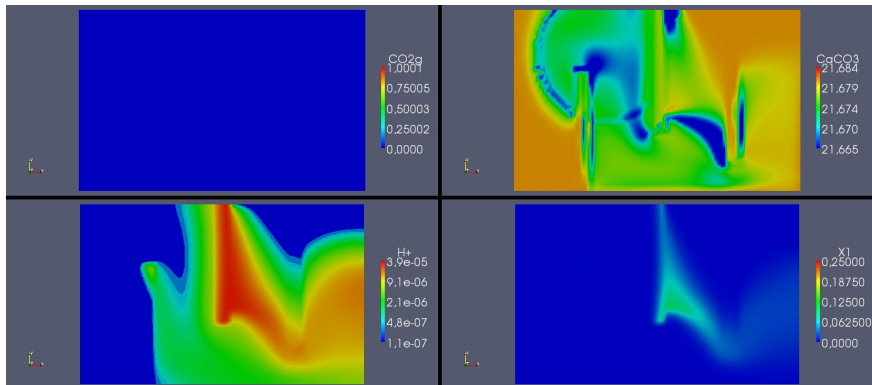
$t = 3600$ years

Evolution of concentrations (Post-doc B. Gueslin)



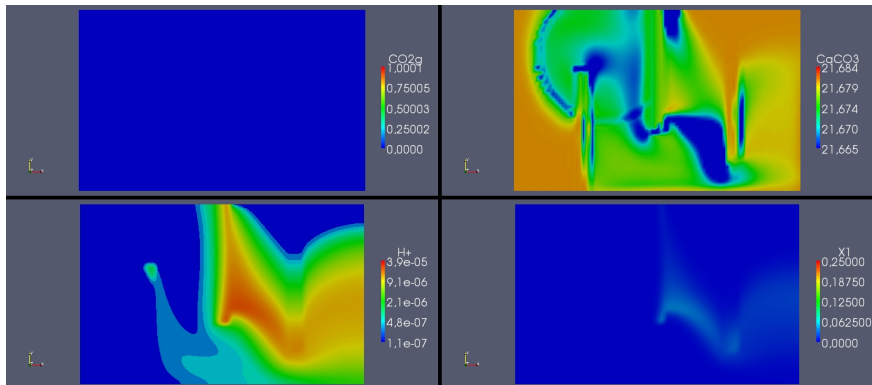
$t = 4000$ years

Evolution of concentrations (Post-doc B. Gueslin)



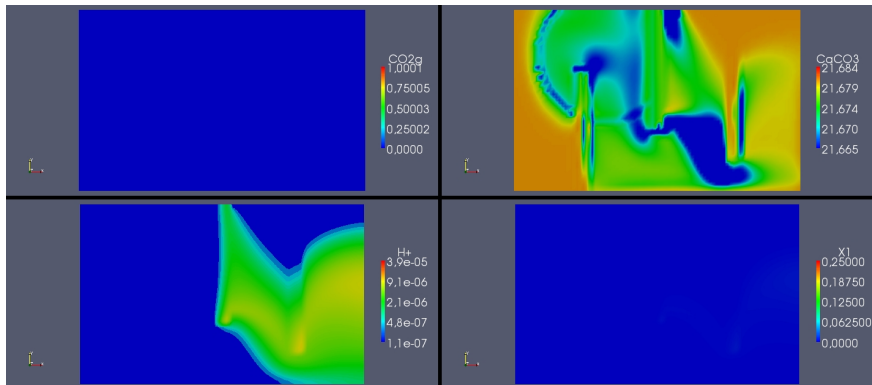
$t = 5000$ years

Evolution of concentrations (Post-doc B. Gueslin)



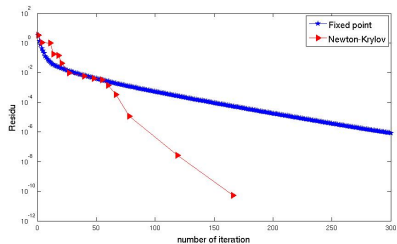
$t = 7000$ years

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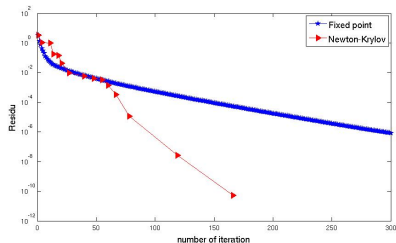


$t = 10000$ years

Newton–Krylov and preconditioner performance



Newton–Krylov and preconditioner performance

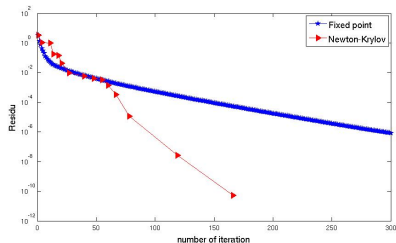


	h		$h/2$		$h/4$		$h/8$	
	NI	LI	NI	LI	NI	LI	NI	LI
No prec.	8	42	8	76	10	105	10	177

Mesh dependence : **adaptive** forcing term

NI : # nonlinear iters, NLI : total # linear iters.

Newton–Krylov and preconditioner performance

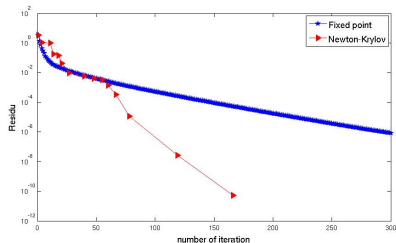


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Elimination	5	15	5	15	5	15	5	15

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Theses : S. Sabit (JE, Andra), V. Vostrikov (B. Amaziane, MK, Maison de la Simulation)

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- Exploring mesh refinement strategies
- Two phase flow with chemistry

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